The ten-membered heterocyclic ring asssumes a distorted chair conformation giving this conformation to the molecule as a whole. The ring is approximately symmetrical with respect to a pseudo-twofold axis through S 1 and C 7 and to a pseudo-mirror plane passing through the middle of bond $\mathrm{C} 1-\mathrm{C} 6$. The rotational symmetry is dominant: the asymmetry parameters (Duax \& Norton, 1975) are $\Delta_{2}=13.8$ and $\Delta_{m}=29.1$. The symmetry distortion mainly results from the large difference of the torsion angles of the pair $\mathrm{S} 1-\mathrm{N} 1-\mathrm{Cl}-\mathrm{C} 6$ and $\mathrm{C} 1-\mathrm{C} 6-\mathrm{O} 3-$ C7. Two fused benzene rings are nearly planar with the maximum deviations from the least-squares planes 0.009 (4) for C5 and 0.013 (4) $\AA$ for C9. The dihedral angle of benzene-ring planes is $5.13(10)^{\circ}$.
$\mathrm{N} 1-\mathrm{H} 100$ and $\mathrm{N} 2-\mathrm{H} 200$ bonds are intraannular with torsion angles $\mathrm{C} 6-\mathrm{C} 1-\mathrm{N} 1-\mathrm{H} 100$ of $-31.3(2.2)$ and $\mathrm{C} 8-\mathrm{C} 13-\mathrm{N} 2-\mathrm{H} 200$ of $-28.6(2.6)^{\circ}$. The shortness of the $\mathrm{N} 1 \cdots \mathrm{O} 3$ and $\mathrm{N} 2 \cdots \mathrm{O} 4$ distances [ 2.755 (2) and 2.770 (2) $\AA$, respectively] suggests there are two intramolecular hydrogen bonds, the $\mathrm{N} 1-\mathrm{H} 100 \cdots \mathrm{O} 3$ and $\mathrm{N} 2-\mathrm{H} 200 \cdots \mathrm{O} 4$
angles being 103.5 (2.5) and $95.6(2.4)^{\circ}$, respectively. One of the two N atoms forms an intermolecular hydrogen bond $\mathrm{N} 1-\mathrm{H} 100 \cdots \mathrm{O}$, where O 1 belongs to the molecule related by $x+1, y, z$. The $\mathrm{N} 1 \cdots \mathrm{O} 1$ distance is 2.859 (3) $\AA$ and the $\mathrm{N} 1-\mathrm{H} 100 \cdots \mathrm{O} 1$ angle is $148.1(2.8)^{\circ}$. Thus the crystal structure is built from chains of molecules linked by hydrogen bonds; these chains run in the [100] direction.

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# Structure of (22S)-3 $\beta$-Acetoxy-20-(3-isopropylisoxazolin-5-yl)-4,4,14 $\alpha$ -trimethylpregn-8(9)-ene 

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#### Abstract

C}_{32} \mathrm{H}_{51} \mathrm{NO}_{3}, \quad M_{r}=497 \cdot 7\), orthorhombic, $P 2_{12} 2_{1}, \quad a=7.577$ (2), $\quad b=10.510$ (2), $\quad c=$ 35.399 (7) $\AA, \quad V=2819(1) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.173 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda(\mathrm{Mo} \mathrm{K} \alpha)=0.71073 \AA, \quad \mu=$ $0.69 \mathrm{~cm}^{-1}, F(000)=1096, T=153 \mathrm{~K}, R=0.0497$ for 2235 observed reflections. The compound investigated is found to be a $(22 S)$-epimer.

Introduction. Steroids having a lanostane skeleton and oxygen function in their side chain at C22 exhibit some useful biological properties (Kamernitzky \& Reshetova, 1977). Isolation of such compounds from natural products and their


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synthesis may be of considerable interest. We have synthesized two isomers [(22R) and (11S)] of $3 \beta$ -acetoxy-20-(3-isopropylisoxazolin-5-yl)-4,4,14 $\alpha$-tri-methylpregn-8(9)-ene (1). Biological activity of such steroid derivatives depends on the configuration of the C22 centre. Therefore the determination of chirality at the C 22 atom in one epimer is of primary importance especially taking into account that further transformations proceed without inversion at the C22 atom.

Experimental. The title compound (1) was prepared from $3 \beta$-acetoxy-25,26,27-trinorlanosta-8,22-diene (Poyser, Hirtzbach \& Ourisson, 1974; Akhrem, Khripach, Litvinovskaya \& Baranovsky, 1989) via (c) 1992 International Union of Crystallography

Table 1. Atomic coordinates ( $\times 10^{4}$ ) and equivalent isotropic displacement coefficients $\left(\AA^{2} \times 10^{3}\right)$
$U_{\mathrm{eq}}$ is defined as one third of the trace of the orthogonalized $U_{i j}$ tensor.

|  | $x$ | $y$ | $z$ | $U_{\text {eq }}$ |
| :---: | :---: | :---: | :---: | :---: |
| C1 | -4261 (6) | -8367 (5) | -1423 (1) | 29 (1) |
| C2 | - 3941 (6) | -8145 (5) | - 1001 (1) | 33 (1) |
| C3 | -2270 (6) | -8802 (4) | -875 (1) | 30 (1) |
| C4 | -602 (6) | -8423 (5) | -1092 (1) | 32 (1) |
| C5 | - 1013 (5) | -8536 (4) | -1524 (1) | 25 (1) |
| C6 | 563 (5) | -8180 (5) | -1770 (1) | 33 (1) |
| C7 | 269 (5) | -8554 (4) | -2175 (1) | 28 (1) |
| C8 | -1607 (5) | -8453 (4) | -2315 (1) | 20 (1) |
| C9 | - 2964 (5) | -8232 (4) | -2085 (1) | 21 (1) |
| C10 | -2737 (5) | - 7883 (4) | -1667 (1) | 22 (1) |
| C11 | -4854 (5) | -8180 (4) | -2236 (1) | 24 (1) |
| C 12 | - 5133 (5) | -8417 (4) | -2662 (1) | 26 (1) |
| C13 | -3518 (5) | -8038 (4) | -2890 (1) | 20 (1) |
| C14 | -1910 (5) | -8785 (4) | -2723 (1) | 21 (1) |
| C15 | -446 (5) | -8440 (4) | -3008 (1) | 28 (1) |
| C16 | -1400 (5) | -8454 (5) | -3395 (1) | 29 (1) |
| C17 | - 3431 (5) | -8446 (4) | -3313 (1) | 23 (1) |
| C18 | -3232 (6) | -6584 (4) | -2859 (1) | 29 (1) |
| C19 | -2675 (7) | -6409 (4) | -1655 (1) | 33 (1) |
| C20 | -4464 (6) | -7686 (4) | -3613 (1) | 28 (1) |
| C21 | -6452 (6) | -7632 (5) | -3531 (1) | 33 (1) |
| C22 | -4096 (6) | -8276 (4) | -4000 (1) | 31 (1) |
| C23 | - 5102 (6) | -7703 (5) | -4336 (1) | 36 (1) |
| C24 | -6386 (6) | -8751 (4) | -4418 (1) | 31 (1) |
| C25 | --7878 (7) | -8668 (5) | -4700 (1) | 36 (2) |
| C26 | -8099 (7) | -9884 (5) | -4924 (2) | 49 (2) |
| C27 | -9581 (7) | -8324 (6) | -4490 (2) | 50 (2) |
| C28 | -2042 (6) | -9357 (4) | -215 (1) | 33 (1) |
| C29 | -1821 (7) | -8782 (5) | 172 (1) | 45 (2) |
| C30 | -2141 (6) | - 10244 (4) | -2741 (1) | 30 (1) |
| C41 | 850 (7) | -9396 (6) | -991 (1) | 48 (2) |
| C42 | 54 (7) | -7110 (5) | -977 (1) | 42 (2) |
| 03 | -2013 (4) | -8465 (3) | -478 (1) | 35 (1) |
| 022 | -4629 (4) | -9612 (3) | - 3993 (1) | 37 (1) |
| O28 | -2253 (6) | - 10468 (3) | -273 (1) | 57 (1) |
| N24 | -6110(5) | -9778 (4) | -4240 (1) | 36 (1) |
|  |  |  |  |  |
| Perspective view of compound (1). |  |  |  |  |

1,3-dipolar cycloaddition of isopropylnitrile oxide. Suitable crystals of the individual epimer (m.p. $522-523 \mathrm{~K}$ ) were prepared by crystallization from EtOAc.
The X-ray intensity data from a colourless $0.30 \times$ $0.35 \times 0.85 \mathrm{~mm}$ crystal were collected with an automated Syntex $P 2_{1}$ four-circle diffractometer at 153 K using graphite-monochromated Mo $K \alpha$ radiation. The unit-cell parameters were determined on the basis of accurately measured positions of 24 reflections in the $13<\theta<14^{\circ}$ range. The intensities of 3733 reflections $(0 \leq h \leq 9, \quad 0 \leq k \leq 13, \quad 0 \leq l \leq 46)$ were measured using $\theta / 2 \theta$ scans (background/peak time ratio 0.25 ) in the $\theta$ range from 2.0 to $30.0^{\circ}$ with a variable scan speed ( $3 \cdot 0-30 \cdot 0^{\circ} \mathrm{min}^{-1}$ in $\omega$ ). The intensities of two standard reflections (measured every 98 reflections) did not show any significant changes $[\delta(I)<3 \%$ ].

Table 2. Torsion angles ( ${ }^{\circ}$ )

| C10 | Cl | C2 | C3 | - 58.8 | C9 | C8 | Cl 4 | Cl 5 | 152.8 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Cl | C2 | C3 | C4 | 56.5 | C9 | C8 | C14 | C30 | -87.5 |
| C1 | C2 | C3 | O3 | 176.2 | C 12 | Cl 3 | C14 | C8 | $-60.8$ |
| C2 | C3 | C4 | C5 | -50.2 | Cl 2 | C13 | C14 | C 15 | 173.4 |
| C2 | C3 | C4 | C41 | - 167.0 | C12 | C 13 | C14 | C30 | 59.5 |
| C2 | C3 | C4 | C42 | 74.9 | C 17 | Cl 3 | C14 | C8 | [74.9 |
| O3 | C3 | C4 | C5 | $-169.2$ | C17 | Cl 3 | C14 | C15 | 49.1 |
| O3 | C3 | C4 | C41 | 74.0 | C 17 | Cl 3 | C14 | C30 | -64.8 |
| O3 | C3 | C4 | C42 | -44.1 | Cl 8 | Cl 3 | C14 | C8 | 58.7 |
| C3 | C4 | C5 | C6 | $-180.0$ | C18 | Cl 3 | C14 | C15 | $-67.1$ |
| C3 | C4 | C5 | C10 | 49.5 | C18 | Cl 3 | Cl 4 | C30 | 179.0 |
| C41 | C4 | C5 | C6 | -63.8 | C8 | C14 | Cl 5 | C16 | $-160.7$ |
| C41 | C4 | C5 | C10 | $165 \cdot 6$ | Cl 3 | Cl 4 | C15 | C16 | -39.1 |
| C42 | C4 | C5 | C6 | 55.8 | C30 | Cl 4 | C15 | C16 | 79.6 |
| C42 | C4 | C5 | C10 | $-74.7$ | C14 | Cl 5 | C16 | C17 | 14.7 |
| C4 | C5 | C6 | C7 | 168.1 | C 12 | Cl 3 | C17 | C 16 | -155.8 |
| C10 | C5 | C6 | C7 | - 58.6 | C 12 | Cl 3 | C17 | C20 | 79.3 |
| C5 | C6 | C7 | C8 | 34.5 | C14 | Cl 3 | C17 | C16 | - 39.4 |
| C6 | C7 | C8 | C9 | -9.6 | C14 | Cl 3 | C17 | C20 | -164.3 |
| C6 | C7 | C8 | C14 | 178.2 | C18 | Cl 3 | C17 | C16 | 77.9 |
| C7 | C8 | C9 | C10 | 8.5 | C18 | Cl 3 | C17 | C20 | -47.0 |
| C7 | C8 | C9 | C11 | $-178.0$ | C 15 | Cl 6 | C17 | C13 | 15.7 |
| C14 | C8 | C9 | C10 | $-179.7$ | C15 | C16 | C17 | C20 | 145.5 |
| C14 | C8 | C9 | C11 | -6.1 | C13 | C 17 | C20 | C21 | - 57.9 |
| C2 | Cl | C10 | C5 | 55.6 | C13 | C 17 | C 20 | C 22 | 178.5 |
| C2 | Cl | C10 | C9 | 173.9 | C16 | Cl 7 | C20 | C21 | -178.3 |
| C2 | Cl | C10 | C19 | -68.8 | C16 | C 17 | C20 | C22 | $58 \cdot 1$ |
| C4 | C5 | C10 | Cl | -52.8 | Cl 17 | C20 | C22 | C 23 | 176.2 |
| C4 | C5 | C10 | C9 | -174.2 | Cl 7 | C20 | C22 | O 22 | 58.4 |
| C4 | C5 | C10 | C19 | 69.1 | C21 | C 20 | C22 | C23 | 51.7 |
| C6 | C5 | C10 | Cl | 176.0 | C21 | C20 | C22 | O 22 | -66.1 |
| C6 | C5 | C10 | C9 | 54.6 | C20 | C22 | C23 | C24 | -108.0 |
| C6 | C5 | C10 | C19 | -62.1 | O 22 | C22 | C23 | C24 | 12.2 |
| C8 | C9 | C10 | Cl | - 148.8 | C22 | C23 | C24 | C25 | $173 \cdot 1$ |
| C8 | C9 | C10 | C5 | -30.5 | C22 | C23 | C24 | N24 | $-9.2$ |
| C8 | C9 | C10 | C19 | 91.5 | C23 | C24 | C25 | C26 | 139.6 |
| Cl 1 | C9 | C10 | Cl | 37.4 | C23 | C24 | C25 | C 27 | -96.9 |
| Cl 1 | C9 | C10 | C5 | 155.6 | N24 | C24 | C25 | C26 | -37.9 |
| Cl 1 | C9 | C10 | C19 | -82.4 | N24 | C24 | C25 | C27 | 85.5 |
| C8 | C9 | C11 | C 12 | 1.2 | C2 | C3 | O3 | C28 | 116.9 |
| C10 | C9 | Cl1 | Cl 2 | $175 \cdot 1$ | C. 4 | C3 | O3 | C28 | - 118.8 |
| C9 | C11 | C12 | Cl 3 | -27.8 | C29 | C28 | O3 | C3 | $-177.7$ |
| Cl 1 | C12 | C13 | C14 | 55.5 | O28 | C28 | O3 | C3 | 0.7 |
| Cl 1 | Cl 2 | C13 | C17 | 168.7 | C20 | C22 | O 22 | N24 | 112.8 |
| Cl 1 | Cl 2 | Cl 3 | C18 | -65.0 | C23 | C22 | O 22 | N24 | -12.3 |
| C7 | C8 | C14 | C13 | $-151 \cdot 1$ | C23 | C24 | N24 | O 22 | 1.9 |
| C7 | C8 | C14 | Cl 5 | -34.9 | C25 | C24 | N24 | O 22 | 179.7 |
| C7 | C8 | C14 | C30 | 84.8 | C22 | O 22 | N24 | C24 | 6.9 |
| C9 | C8 |  |  |  |  |  |  |  |  |

All calculations were performed with an IBM-PC/AT computer using SHELXTL-Plus (PC Version; Sheldrick, 1989). The intensities of 2235 independent observed $[I>3 \sigma(I)$ ] reflections corrected for Lorentz and polarization factors were used. The structure was solved using direct methods and refined by full-matrix least squares in anisotropic approximation for non- H atoms. H atoms were placed in calculated positions and refined isotropically with a fixed $U_{\text {iso }}=0.08 \AA^{2}$. The final refinement of 289 parameters converged to $R=0.0497, w R=$ $0.0498 \quad\left[w^{-1}=\sigma^{2}(F)+99 \cdot 000 F^{2}\right], \quad S=1 \cdot 17$, mean $(\Delta / \sigma)=0.170,0.23>\Delta \rho>-0.19 \mathrm{e} \AA^{-3}$.

Discussion. Atomic coordinates and equivalent isotropic thermal parameters are given in Table 1.* The

[^1]molecular structure with atom numbering is shown in Fig. 1. Bond lengths and angles are unexceptional (Allen, Kennard, Watson, Brammer, Orpen \& Taylor, 1987; Duax \& Norton, 1975).

The determination of absolute configuration of the C22 chiral centre, which turned out to be $S$, was based on the known chirality of the steroid framework. The main torsion angles, which characterize the molecular conformation, are given in Table 2. The steroid framework conformation coincides with a standard for $\Delta^{8(9)}$-pregnanes (Duax \& Norton, 1975), viz. a chair for the $A$ ring, a $5 \alpha$-sofa for the $B$ ring, a $13 \beta$-sofa for the $C$ ring and a $14 \alpha$-envelope for the $D$ ring. The orientation of the side chain at C17, characterized by the torsion angle C16 $\mathrm{C} 17 \mathrm{C} 20 \mathrm{C} 21=-178 \cdot 3^{\circ}$, may be considered as sterically least strained. The oxazoline ring is in fact planar. The orientation of the $3 \beta$-acetoxy group
[with the carbonyl O and $\mathrm{H}(\mathrm{C} 3)$ atoms disposed close to each other] is similar to that usually observed in most of the analogous steroid derivatives (Duax \& Norton, 1975).

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# Structure of Propyl 3-Acetyl-4-[2-hydroxy-3-(isopropylamino)propoxy|carbanilate Hydrochloride 

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#### Abstract

C}_{18} \mathrm{H}_{29} \mathrm{~N}_{2} \mathrm{O}_{5}^{+} . \mathrm{Cl}^{-}, M_{r}=388.9\), monoclinic, $C 2 / c, a=42.014$ (8), $b=5.054$ (1), $c=30.023$ (5) $\AA$, $\beta=139.84$ (4) ${ }^{\circ}, \quad V=4111.4$ (6) $\AA^{3}, \quad Z=8, \quad D_{m}=$ 1.25 (1), $\quad D_{x}=1.256 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{Cu} \mathrm{K} \mathrm{\alpha})=$ $1.54178 \AA, \quad \mu=0.94 \mathrm{~mm}^{-1}, \quad F(000)=1664, \quad T=$ $293 \mathrm{~K}, R=0.052$ for 1974 unique observed reflections. The structure consists of discrete cations connected by hydrogen-bonded chloride anions. The arrangement of the aryloxy and 2-hydroxyl groups around the conformationally flexible $\mathrm{OCH}_{2}-$ $\mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{2}$ bond of the oxypropanolamine side chain is gauche. While the lone-pair electrons of the ethereal oxygen are definitely delocalized through the adjacent phenyl ring, the results do not reveal any significant conjugation of the carbamate and acetyl groups with the aromatic system.


Introduction. This work is part of a more general study on conformational properties of a new series of $\beta$-adrenoceptor blocking drugs belonging to the aryloxypropanolamine family. Another interest in the present structure results from the observation that while the title compound is a highly potent $\beta$-adrenoceptor antagonist, its positional isomer, having interchanged the acetyl and carbamate functions on the aromatic ring, has been found to be totally inactive (Csöllei, Račanská, Švec \& Kettmann, 1991). The inactivity of the latter was attributed to the positive molecular electrostatic potential (MEP) generated by the $\pi$-deficient phenyl ring as a result of through-conjugation between the ethereal oxygen and the acceptor acetyl group (Kettmann, 1991). Consequently, the structure deter-


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[^1]:    * Lists of structure factors, anisotropic thermal parameters, full bond lengths and angles, and H -atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54394 (20 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

